



DOWEX SBR-P

A High Efficiency, Strong Base Anion Exchange Resin for Water Demineralization Applications

Product	Type	Matrix	Functional group
DOWEX* SBR-P	Type 1 strong base anion	Styrene-DVB, gel	Quaternary amine

Guaranteed Sales Specifications		Cl ⁻ form
Total exchange capacity, min.	eq/l	1.3
	kgr/ft ³ as CaCO ₃	28.4
Bead size distribution range [†] 0.3 mm - 1.2 mm, min. (50 mesh - 16 mesh)	%	90

Typical Physical and Chemical Properties		Cl ⁻ form
Water content	%	50 - 56
Whole uncracked beads	%	90 - 100
Total swelling (Cl ⁻ → OH ⁻)	%	20
Particle density	g/ml	1.08
Shipping weight	g/l lbs/ft ³	690 43

Recommended Operating Conditions	
Maximum operating temperature:	
OH ⁻ form	60°C (140°F)
Cl ⁻ form	100°C (212°F)
pH range	0-14
Bed depth, min.	800 mm (2.6 ft)
Flow rates:	
Service/fast rinse	5-50 m/h (2-20 gpm/ft ²)
Backwash	See figure 1
Co-current regeneration/displacement rinse	1-10 m/h (0.4-4 gpm/ft ²)
Total rinse requirement	3-6 Bed volumes
Regenerant:	
Type	2-5% NaOH
Temperature	Ambient or up to 50°C (122°F) for silica removal
Load of organic matter, max.	3 g KMnO ₄ /l

[†]For additional particle size information, please refer to the Particle Size Distribution Cross Reference Chart (Form No. 177-01775/CH 171-476-E).

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DOWEX Ion Exchange Resins

For more information about DOWEX resins, call Dow Liquid Separations business:
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Typical properties and applications:

DOWEX* SBR-P type 1 strong base anion resin has excellent kinetics, very good regeneration efficiency and physical and chemical stability. The porous bead is made by a special process giving enhanced

resistance to organics and fast equilibrium rates. Mainly used for demineralization of water, extraction of heavy metals and the recovery of precious metals in the form of complex anions.

Packaging

25 liter bags or 5 cubic feet fiber drums.

Figure 1. Backwash Expansion Data

Temperature = 25° C (77° F)

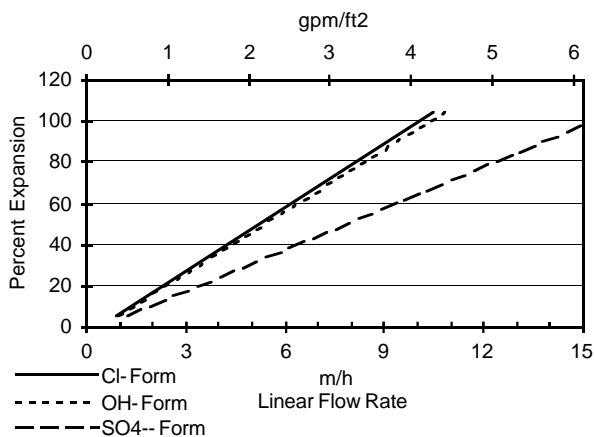
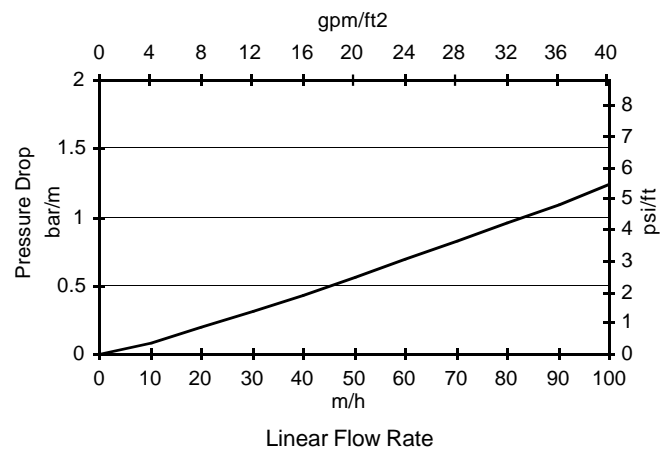


Figure 2. Pressure Drop Data

Temperature = 20° C (68° F)



For other temperatures use:

$$F_T = F_{77°F} [1 + 0.008 (T_{°F} - 77)], \text{ where } F \equiv \text{gpm/ft}^2$$

$$F_T = F_{25°C} [1 + 0.008 (1.8T_{°C} - 45)], \text{ where } F \equiv \text{m/h}$$

For other temperatures use:

$$P_T = P_{20°C} / (0.026 T_{°C} + 0.48), \text{ where } P \equiv \text{bar/m}$$

$$P_T = P_{68°F} / (0.014 T_{°F} + 0.05), \text{ where } P \equiv \text{psi/ft}$$

Warning: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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